

## *Intermediate Chemistry Notes*







**Intermediate Chemistry Notes**

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Chemical Processing Definition. The term "intermediate" means something different in the chemical industry, referring to a stable product of a chemical reaction that is then used as a starting material for another reaction. For example, benzene and propylene may be used to make the intermediate cumene. Cumene is then used to make phenol and acetone.

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But in the equation of the process 1 mole of nitrogen require 3 mole of hydrogen. Therefore the provided moles of nitrogen i.e. 1.79 require  $1.79 \times 3$  moles of hydrogen i.e. 5.37 moles although 25 moles of  $H_2$  are provided but when nitrogen is consumed the reaction will be stopped and the remaining hydrogen is useless for the reaction so in this problem  $N_2$  is a limiting reactant by which we can ...

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